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## **Plant Physiology (Mineral Nutrition) - Part 1**

### **Objectives**

After going through this lesson, the learners will be able to understand the following:

- Describe the methods to study the mineral requirements of plants.
- Delineate the essential mineral elements for plants.
- Discuss the criteria for essentiality of an element for a plant.
- Discuss the role of macro- and micro-nutrients for plants.
- Explain the deficiency symptoms of essential elements for plants.
- Define the toxicity of micronutrients for plants.

### **Content Outline**

- Introduction
- Methods to Study the Mineral Requirements of Plants
- Essential Mineral Elements
- Summary

### **Introduction**

The basic needs of all living organisms are essentially the same. They require macromolecules, such as carbohydrates, proteins and fats, and water and minerals for their growth and development. This chapter focuses mainly on inorganic plant nutrition, wherein you will study the methods to identify elements essential to growth and development of plants and the criteria for establishing the essentiality. You will also study the role of the essential elements, their major deficiency symptoms and the mechanism of absorption of these essential elements. The chapter also introduces you briefly to the significance and the mechanism of biological nitrogen fixation.

### **Methods to Study the Mineral Requirements of Plants**

In 1860, Julius von Sachs, a prominent German botanist, demonstrated, for the first time, that plants could be grown to maturity in a defined nutrient solution in complete absence of soil. This technique of growing plants in a nutrient solution is known as hydroponics. Since then, a number of improvised methods have been employed to try and determine the mineral nutrients essential for plants. The essence of all these methods involves the culture of plants

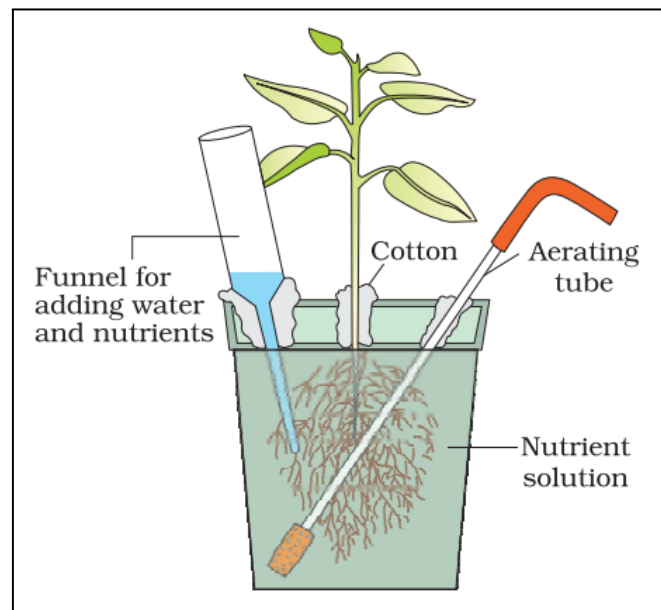
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in a soil-free, defined mineral solution. These methods require purified water and mineral nutrient salts.

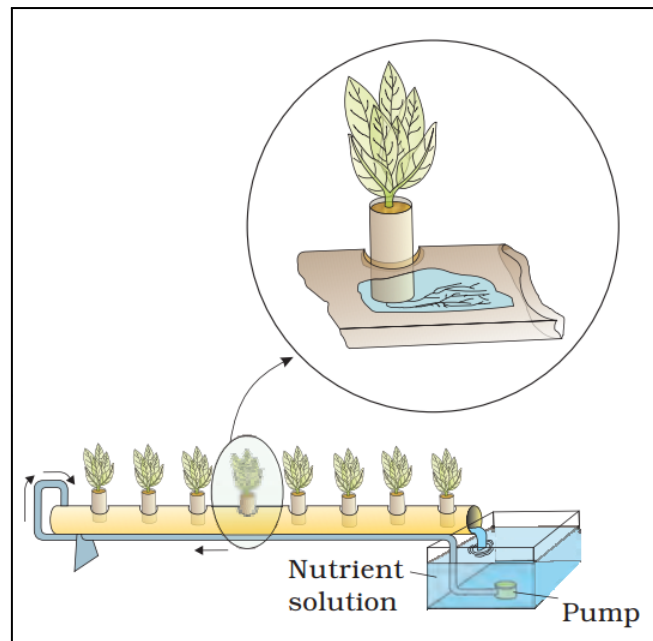
**Point to Ponder**

- Can you explain why mineral nutrition is so essential for plants?

After a series of experiments in which the roots of the plants were immersed in nutrient solutions and wherein an element was added/ removed or given in varied concentration, a mineral solution suitable for the plant growth was obtained. By this method, essential elements were identified and their deficiency symptoms discovered. Hydroponics has been successfully employed as a technique for the commercial production of vegetables such as tomato, seedless cucumber and lettuce. It must be emphasised that the nutrient solutions must be adequately aerated to obtain the optimum growth. What would happen if solutions were poorly aerated? Diagrammatic views of the hydroponic technique are given in Figures 1 and 2.



**Figure 1:** Diagram of a typical set-up for nutrient solution culture.



**Figure 2:** Hydroponic plant production.

Plants are grown in a tube or trough placed on a slight incline. A pump circulates a nutrient solution from a reservoir to the elevated end of the tube. The solution flows down the tube and returns to the reservoir due to gravity. Inset shows a plant whose roots are continuously bathed in aerated nutrient solution. The arrows indicate the direction of the flow.

### **Essential Mineral Elements**

Most of the minerals present in soil can enter plants through roots. In fact, more than sixty elements of the 105 discovered so far are found in different plants. Some plant species accumulate selenium, some others gold, while some plants growing near nuclear test sites take up radioactive strontium. There are techniques that are able to detect the minerals even at a very low concentration ( $10^{-8}$  g/mL). The elements particularly C, H and O construct the plant body by entering into the constitution of cell wall and protoplasm. They are, therefore, referred to as framework elements. Besides, these (C, H and O) N, P and S also enter in the constitution of protoplasm. They are described as protoplasmic elements. Nutrients must be available not only in sufficient amounts but also in appropriate ratios. The question is, whether all the diverse mineral elements present in a plant, for example, gold and selenium as mentioned above, are really necessary for plants?

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### Point to Ponder

- How do we decide that in terms of mineral nutrition what is essential for plants and what is not?
- Do all the diverse mineral elements present in a plant, for example, gold and selenium are really necessary for plants?

### Criteria for Essentiality

The criteria for essentiality of an element are given below:

- a) The element must be absolutely necessary for supporting normal growth and reproduction. In the absence of the element the plants do not complete their life cycle or set the seeds.
- b) The requirement of the element must be specific and not replaceable by another element. In other words, deficiency of any one element cannot be met by supplying some other element.
- c) The element must be directly involved in the metabolism of the plant.

Plants take up essential elements from the soil through their roots and from the air (mainly consisting of nitrogen and oxygen) through their leaves. Nutrient uptake in the soil is achieved by cation exchange, wherein root hairs pump hydrogen ions ( $H^+$ ) into the soil through proton pumps. These hydrogen ions displace cations attached to negatively charged soil particles so that the cations are available for uptake by the root. In the leaves, stomata open to take in carbon dioxide and expel oxygen. The carbon dioxide molecules are used as the carbon source in photosynthesis.

Based upon the above criteria only a few elements have been found to be absolutely essential for plant growth and metabolism. These elements are further divided into two broad categories based on their quantitative requirements.

- I. Macronutrients
- II. Micronutrients

The **Macronutrients** are generally present in plant tissues in large amounts (in excess of 10 mmole  $Kg^{-1}$  of dry matter). The macronutrients include carbon, hydrogen, oxygen, nitrogen, phosphorus, sulphur, potassium, calcium and magnesium. Of these, carbon, hydrogen and

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oxygen are mainly obtained from CO<sub>2</sub> and H<sub>2</sub>O, while the others are absorbed from the soil as mineral nutrition.

The **Micronutrients** or trace elements, are needed in very small amounts (less than 10 mmole Kg<sup>-1</sup> of dry matter). These include iron, manganese, copper, molybdenum, zinc, boron, chlorine and nickel. In addition to the 17 essential elements named above, there are some beneficial elements such as sodium, silicon, cobalt and selenium. They are required by higher plants. Essential elements can also be grouped into four broad categories on the basis of their diverse functions. These categories are:

- I. Essential elements as components of biomolecules and hence structural elements of cells (e.g., carbon, hydrogen, oxygen and nitrogen).
- II. Essential elements that are components of energy-related chemical compounds in plants (e.g., magnesium in chlorophyll and phosphorus in ATP).
- III. Essential elements that activate or inhibit enzymes, for example Mg<sup>2+</sup> is an activator for both ribulose biphosphate carboxylase oxygenase and phosphoenol pyruvate carboxylase, both of which are critical enzymes in photosynthetic carbon fixation; Zn<sup>2+</sup> is an activator of alcohol dehydrogenase and Mo of nitrogenase during nitrogen metabolism.

**Point to Ponder**

- Can you name a few more elements that fall in this category? For this, you will need to recollect some of the biochemical pathways you have studied earlier.

- IV. Some essential elements can alter the osmotic potential of a cell. Potassium plays an important role in the opening and closing of stomata. You may recall the role of minerals as solutes in determining the water potential of a cell.

**Role of Macro and Micro Nutrients**

Essential elements perform several functions. They participate in various metabolic processes in the plant cells such as permeability of cell membrane, maintenance of osmotic concentration of cell sap, electron transport systems, buffering action, enzymatic activity and act as major constituents of macromolecules and co-enzymes. Various minerals present in the

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cell sap in organic or inorganic form maintain the osmotic pressure of the cell. Different cations and anions influence the *pH* of the cell sap. Various forms and functions of essential nutrient elements are given below.

**Nitrogen:** This is the essential nutrient element required by plants in the greatest amount. It is absorbed mainly as  $\text{NO}_3^-$  though some are also taken up as  $\text{NO}_2^-$  or  $\text{NH}_4^+$ . Nitrogen is required by all parts of a plant, particularly the meristematic tissues and the metabolically active cells. Nitrogen is one of the major constituents of proteins, nucleic acids, vitamins and hormones. Nitrogen is also present in various coenzymes and ATP etc.

**Point to Ponder**

**What is the reason behind the fact that despite an abundant supply of nitrogen in the earth's atmosphere,  $\text{N}_2$  is unavailable for use by most organisms?**

There is an abundant supply of nitrogen in the earth's atmosphere —  $\text{N}_2$  gas comprises nearly 79% of air. However,  $\text{N}_2$  is unavailable for use by most organisms because there is a triple bond between the two nitrogen atoms in the molecule, making it almost inert. In order for nitrogen to be used for growth it must be “fixed” (combined) in the form of ammonium ( $\text{NH}_4^+$ ) or nitrate ( $\text{NO}_3^-$ ) ions. The weathering of rocks releases these ions so slowly that it has a negligible effect on the availability of fixed nitrogen. Therefore, nitrogen is often the limiting factor for growth and biomass production in all environments where there is a suitable climate and availability of water to support life.

**Phosphorus:** Phosphorus is absorbed by the plants from soil in the form of phosphate ions (either as  $\text{HPO}_2^{4-}$  or  $\text{HPO}_4^{2-}$ ). Phosphorus is a constituent of cell membranes, certain proteins, all nucleic acids and nucleotides, and is required for all phosphorylation reactions. It promotes healthy root growth and fruit ripening by helping translocation of carbohydrates.

**Potassium:** It is absorbed as potassium ion ( $\text{K}^+$ ). In plants, this is required in more abundant quantities in the meristematic tissues, buds, leaves and root tips. Potassium helps to maintain an anion-cation balance in cells and is involved in protein synthesis, opening and closing of stomata, activation of enzymes and in the maintenance of the turgidity of cells.

**Calcium:** Plant absorbs calcium from the soil in the form of calcium ions ( $\text{Ca}^{2+}$ ). Calcium is required by meristematic and differentiating tissues. During cell division it is used in the

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synthesis of cell wall, particularly as calcium pectate in the middle lamella. It is also needed during the formation of mitotic spindles. It accumulates in older leaves. It is involved in the normal functioning of the cell membranes. It activates certain enzymes and plays an important role in regulating metabolic activities.

**Magnesium:** It is absorbed by plants in the form of divalent  $Mg^{2+}$ . It activates the enzymes of respiration, photosynthesis and are involved in the synthesis of DNA and RNA. Magnesium is a constituent of the ring structure of chlorophyll and helps to maintain the ribosome structure. It plays an important role in the metabolism of carbohydrates, lipids and phosphorus.

**Sulphur:** Plants obtain sulphur in the form of sulphate  $SO_4^{2-}$ . Sulphur is present in two amino acids – cysteine and methionine and is the main constituent of several coenzymes, vitamins (thiamine, biotin, Coenzyme A) and ferredoxin.

**Iron:** Plants obtain iron in the form of ferric ions ( $Fe^{3+}$ ). It is required in larger amounts in comparison to other micronutrients. It is an important constituent of proteins involved in the transfer of electrons like ferredoxin and cytochromes. It is reversibly oxidised from  $Fe^{2+}$  to  $Fe^{3+}$  during electron transfer. It activates catalase enzymes, and is essential for the formation of chlorophyll. Iron is a structural component of ferredoxin, flavoproteins, iron porphyrin proteins (Cytochromes, peroxidases, catalases, etc.)

**Manganese:** It is absorbed in the form of manganous ions ( $Mn^{2+}$ ). It activates many enzymes involved in photosynthesis, respiration and nitrogen metabolism. The best defined function of manganese is in the splitting of water to liberate oxygen during photosynthesis. It acts as an activator of enzymes of respiration (malic dehydrogenase and oxalosuccinate decarboxylase) and nitrogen metabolism (nitrite reductase).

**Zinc:** Plants obtain zinc as  $Zn^{2+}$  ions. It activates various enzymes, especially carboxylases. It is also needed in the synthesis of auxin. Copper: It is absorbed as cupric ions ( $Cu^{2+}$ ). It is essential for the overall metabolism in plants. Like iron, it is associated with certain enzymes involved in redox reactions and is reversibly oxidised from  $Cu^+$  to  $Cu^{2+}$ . It is required for the metabolism of phosphorus and carbohydrates.

**Boron:** It is absorbed as  $BO_3^{3-}$  or  $B_4O_7^{2-}$ . Boron is required for uptake and utilisation of  $Ca^{2+}$ , membrane functioning, pollen germination, cell elongation, cell differentiation and carbohydrate translocation. It is also required for flowering, fruiting, and photosynthesis and nitrogen metabolism. It is mobile in the soil, hence, it is prone to leaching. Leaching removes substantial amounts of boron in sandy soil, but little in fine silt or clay soil. Boron's fixation

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to those minerals at high pH can render boron unavailable, while low pH frees the fixed boron, leaving it prone to leaching in wet climates. It precipitates with other minerals in the form of borax in which form it was first used over 400 years ago as a soil supplement.

**Molybdenum:** Plants obtain it in the form of molybdate ions ( $\text{MoO}_4^{2-}$ ). It is a component of several enzymes, including nitrogenase and nitrate reductase both of which participate in nitrogen metabolism. Its most important function is in nitrogen fixation because it is an activator of nitrate reductase. It is required for the synthesis of ascorbic acid.

**Chlorine:** It is absorbed in the form of chloride anion ( $\text{Cl}^-$ ). Along with  $\text{Na}^+$  and  $\text{K}^+$ , it helps in determining the solute concentration and the anion cation balance in cells. It is essential for the water-splitting reaction in photosynthesis, a reaction that leads to oxygen evolution. It is required for photolysis of water during photosynthesis in photosystem II.

### **Deficiency Symptoms of Essential Elements**

Whenever the supply of an essential element becomes limited, plant growth is retarded. The concentration of the essential element below which plant growth is retarded is termed as critical concentration. The element is said to be deficient when present below the critical concentration. Since each element has one or more specific structural or functional role in plants, in the absence of any particular element, plants show certain morphological changes. These morphological changes are indicative of certain element deficiencies and are called deficiency symptoms.

The deficiency symptoms vary from element to element and they disappear when the deficient mineral nutrient is provided to the plant. However, if deprivation continues, it may eventually lead to the death of the plant. The parts of the plants that show the deficiency symptoms also depend on the mobility of the element in the plant. For elements that are actively mobilised within the plants and exported to young developing tissues, the deficiency symptoms tend to appear first in the older tissues. For example, the deficiency symptoms of nitrogen, potassium and magnesium are visible first in the senescent leaves. In the older leaves, biomolecules containing these elements are broken down, making these elements available for mobilising to younger leaves.



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**Point to Ponder****What is plant senescence?**

Plant senescence is the process of aging in plants. Plants have both stress-induced and age-related developmental aging. Chlorophyll degradation during leaf senescence reveals the carotenoids, and is the cause of autumn leaf colour in deciduous trees. Leaf senescence has the important function of recycling nutrients, mostly nitrogen, to growing and storage organs of the plant. Unlike animals, plants continually form new organs and older organs undergo a highly regulated senescence program to maximize nutrient export.

The deficiency symptoms tend to appear first in the young tissues whenever the elements are relatively immobile and are not transported out of the mature organs, for example, elements like calcium is a part of the structural component of the cell and hence are not easily released. This aspect of mineral nutrition of plants is of a great significance and importance to agriculture and horticulture. The kind of deficiency symptoms shown in plants include chlorosis, necrosis, stunted plant growth, premature fall of leaves and buds, and inhibition of cell division. Chlorosis is the loss of chlorophyll leading to yellowing in leaves. This symptom is caused by the deficiency of elements N, K, Mg, S, Fe, Mn, Zn and Mo. Likewise, necrosis, or death of tissue, particularly leaf tissue, is due to the deficiency of Ca, Mg, Cu, K. Lack or low level of N, K, S, Mo causes an inhibition of cell division. Some elements like N, S, Mo delay flowering if their concentration in plants is low. You can see from the above that the deficiency of any element can cause multiple symptoms and that the same symptoms may be caused by the deficiency of one of several different elements. Hence, to identify the deficient element, one has to study all the symptoms developed in all the various parts of the plant and compare them with the available standard tables. We must also be aware that different plants also respond differently to the deficiency of the same element.

**Point to Ponder**

- What is Chlorosis?
- What is Necrosis?
- Why different plants respond differently to the deficiency of the same element.

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## Toxicity of Micronutrients

The requirement of micronutrients is always in low amounts while their moderate decrease causes the deficiency symptoms and a moderate increase causes toxicity. In other words, there is a narrow range of concentration at which the elements are optimum. Any mineral ion concentration in tissues that reduces the dry weight of tissues by about 10 per cent is considered toxic. Such critical concentrations vary widely among different micronutrients. The toxicity symptoms are difficult to identify.

Toxicity levels for any element also vary for different plants. Many times, excess of an element may inhibit the uptake of another element. For example, the prominent symptom of manganese toxicity is the appearance of brown spots surrounded by chlorotic veins. It is important to know that manganese competes with iron and magnesium for uptake and with magnesium for binding with enzymes. Manganese also inhibits calcium translocation in shoot apex. Therefore, excess manganese may, in fact, induce deficiencies of iron, magnesium and calcium. Thus, what appears as symptoms of manganese toxicity may actually be the deficiency symptoms of iron, magnesium and calcium. Minerals like *Cu*, *As*, etc. impart toxic effects on the protoplasm under specific conditions. Can this knowledge be of some importance to a farmer? a gardener? or even for you in your kitchen-garden?

### Point to Ponder

- What is mineral toxicity for plants?
- How can this knowledge of mineral toxicity for plants be helpful for farmers and gardeners?

## Summary

Plants obtain their inorganic nutrients from air, water and soil. Plants absorb a wide variety of mineral elements. Not all the mineral elements that they absorb are required by plants. Out of the more than 105 elements discovered so far, less than 21 are essential and beneficial for normal plant growth and development. The elements required in large quantities are called macronutrients while those required in less quantities or in trace are termed as micronutrients. These elements are either essential constituents of proteins, carbohydrates, fats, nucleic acid etc., and/or take part in various metabolic processes. Deficiency of each of these essential elements may lead to symptoms called deficiency symptoms. Chlorosis, necrosis, stunted growth, impaired cell division, etc., are some prominent deficiency symptoms. Plants absorb

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minerals through roots by either passive or active processes. They are carried to all parts of the organism through xylem along with water transport.